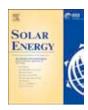
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Optimization of SnO₂-based electron-selective contacts for Si/PEDOT:PSS heterojunction solar cells



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ABSTRACT

Tin oxide (SnO₂) is a potentially excellent electron-selective contact (ESC) for silicon (Si)-based solar cells due to its satisfactory energy band structure and good crystallinity. However, unsatisfactory electron extraction ability and limited surface passivating effect of SnO₂ ESCs will limit the performance of corresponding solar cells. We increase the Fermi level of SnO₂ by doping Ethylene diamine tetraacetic acid (EDTA), which endows EDTA-SnO₂ better electron extraction ability than SnO₂. Moreover, EDTA-SnO₂/SiO_x bilayer ESC prepared by combining a EDTA-SnO₂ layer and a thin silicon oxide (SiO_x) film provides better surface passivation than EDTA-SnO₂ ESC without impairing the charge transport capability markedly. The planar Si/PEDOT:PSS heterojunction solar cells (HSCs) with EDTA-SnO₂/SiO_x bilayer ESCs exhibit a power conversion efficiency (η) of 11.52%, which improves 13.7% in comparison with the η (10.13%) of HSCs with SnO₂ ESCs, mainly caused by the increase in V_{oc} and FF by 18 mV and 5.4% respectively.

1. Introduction

Due to its excellent device stability, high power conversion efficiency (η) and relatively mature industrial system, silicon (Si)-based solar cells occupy more than 90% of the photovoltaic market share. Further technical development direction of Si-based solar cells is to produce more efficient solar cells through simpler preparation process and lower cost. The working principle of the solar cell is that the light absorbing layer absorbs photons to generate electron and hole pairs, then the photogenerated electrons and holes transport under the driving force of Fermi level gradient (Wurfel et al., 2015). Carrier-selective contact (CSC), which extracts just one-type of carrier (electron or hole) from light absorbing layer, enables photogenerated electrons and holes transport to respective electrodes through different paths, thus realizing separation of carriers and utilization of photon energy (Bullock et al., 2018; Gao et al., 2018; Melskens et al., 2018; Yang et al., 2016). The most ideal solar cells should be constructed in the form of double heterojunction (Yablonovitch et al., 1985). In this structure, the narrow bandgap light absorbing layer is sandwiched between two oppositely doped wide bandgap layers that act as CSCs. Commercially available Si-based solar cells realize carrier separation by doping nearsurface regions or preparing Si-based thin films to form CSCs (Bullock et al., 2018; Melskens et al., 2018; Yang et al., 2016). However,

parasitic absorption, Auger recombination and other heavy doping effects caused by doped silicon contacts will limit the η of Si-based solar cells. Meanwhile, it requires high production cost and complicated process to prepare such contacts. Therefore, many researchers are devoted to developing better dopant-free CSCs (Bullock et al., 2018; Melskens et al., 2018; Yang et al., 2016).

When non-heavily doped silicon directly contacting silver (Ag) or aluminum electrodes in Si/PEDOT:PSS heterojunction solar cells (HSCs), high Schottky barrier caused by Fermi level pinning effect in the silicon surface area will hinder the carrier transmission, thus severely damaging the performance of solar cells (Bullock et al., 2018; He et al., 2018; Liu et al., 2017; Melskens et al., 2018; Wan et al., 2017; Yang et al., 2016; Yang et al., 2017; Yoon and Khang, 2018; Zhang et al., 2014). Therefore, preparing electron-selective contacts (ESCs) to promote the separation of photogenerated carriers, passivate the silicon surface and eliminate the Si/metal Schottky barrier are of great significance for improving the η of such HSCs. Metal oxide semiconductors such as titanium oxide (TiO_x) (Cui et al., 2016; Liao et al., 2014; Melskens et al., 2016; Yang et al., 2016), zinc oxide (ZnO) (Robertson, 2000), tin oxide (SnO₂) (Chen et al., 2018; Young et al., 2014) and tantalum oxide (TaOx) (Wan, 2015) have been investigated as ESCs for Si-based solar cells due to their ease of deposition and low cost (Yang et al., 2016). Among them, n-type semiconductor SnO2 has a slightly

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higher conduction band and a very deep valence band comparing to silicon (Ke et al., 2015; Yang et al., 2018). The ideal band structure provides SnO_2 excellent electron transmission and hole blocking capabilities. Moreover, SnO_2 has higher electron mobility than TiO_x and better stability than TiO_x and TiO_x in TiO_x and TiO_x becomes TiO_x because TiO_x because Ti

In this work, we present EDTA-SnO₂/SiO_x bilayer ESCs with better properties comparing to SnO₂ ESCs for Si/PEDOT:PSS HSCs. We improve the Fermi level of SnO₂ by doping trace amount of Ethylene diamine tetraacetic acid (EDTA), which endows EDTA-SnO₂ with better electron extraction ability than SnO₂. Moreover, EDTA-SnO₂/SiO_x bilayer ESC is prepared by combining a EDTA-SnO₂ layer and a thin silicon oxide (SiO_x) film grown on silicon surface. EDTA-SnO₂/SiO_x bilayer ESC provides better surface passivation than EDTA-SnO₂ without impairing the charge transport capability markedly. The properties of SnO₂-based ESCs are evaluated by considering both the contact resistivity and the surface passivation quality on silicon surfaces. The planar Si/PEDOT:PSS HSCs featuring EDTA-SnO₂/SiO_x bilayer ESCs exhibit a higher η (11.52%) comparing to the η (10.13%) of HSCs with SnO₂ ESCs, mainly caused by the increase in V_{oc} and FF by 18 mV and 5.4% respectively.

2. Experiment section

2.1. Materials and devices fabrication

0.04 mg of EDTA (Sigma Aldrich) was dissolved in 1 ml of water, and 15 wt% $\rm SnO_2$ solution (Alfa Aesar) was diluted to 2.5 wt%. Then the two solutions were mixed in equal volume to prepare EDTA- $\rm SnO_2$ solution. The concentration of solution was adjusted by dilution or heating evaporation. PEDOT:PSS (Heraeus, PH1000) solution was doped with 7 wt% ethylene glycol (Innochem) and 0.1 wt% TX-100 (Alfa). The silicon wafer used is (1 0 0)-oriented n-type (1–3 Ω -cm, 270 μ m, Tian Jin upward technology development co., ltd.).

The silicon wafer was ultrasonically cleaned with acetone, ethanol and deionized water for $10\,\mathrm{min}$ respectively, and then soaked in hydrofluoric acid solution (13.3 wt%) for 6 min to remove the SiO_x . After that, HSCs without SnO_2 ESCs was directly plated with Ag electrode at the rear side; for HSCs with SnO_2 or EDTA-SnO $_2$ Solution was spin-coated at 3500 rpm for 20 s at the rear side, then

the solvent was removed by heating (60 °C, 30 min), followed by plating Ag rear electrode; for HSCs with EDTA-SnO₂/SiO_x bilayer ESCs, SiO_x was prepared through ultraviolet (UV)/O₃ photo-oxidation treatment for 4 min by UV ozone cleaning system (Moldovan et al., 2015), and then EDTA-SnO₂ was spin-coated in the same way as mentioned above, followed by plating Ag rear electrode. Finally, PEDOT:PSS solution was spin-coated on the silicon wafer at 5000 rpm for 50 s and then annealed at 120 °C for 15 min, followed by plating front gate Ag electrode.

2.2. Device characterization

The light *J-V* curves of the solar cells were measured using a Keithley 2400 source under ambient conditions at room temperature. The light source was calibrated by a standard silicon solar cell to match AM 1.5 G. The contact resistivity of the rear structures were measured using a Keithley 2400 source. The external quantum efficiency (EQE) of HSC was tested by Zolix SCS100 full function solar cell quantum efficiency test system. XPS was performed by ESCALAB 250Xi, Thermofisher Scientific. Kelvin probe force microscopy (KPFM) was carried out on Aglient Keysight AFM-5500. The minority carrier lifetime was measured by MDPspot system from Freiberg Instruments.

3. Results and discussion

It is reported that EDTA can modify the properties of transition metal oxides by providing its lone-pair electrons to the vacant d-orbital of the transition metal atoms (Li et al., 2017). Moreover, doping of EDTA in SnO₂ can lead to similar result (Yang et al., 2018). We doped trace amount of EDTA in SnO₂ to modify the electron extraction ability of SnO2. Then, we spin-coated SnO2 and EDTA-SnO2 on silicon wafer respectively to test the reaction between EDTA and SnO2 by XPS test, the results were shown in Fig. 1. After doping of EDTA, the XPS spectra of EDTA-SnO₂ shows an additional little peak located at approximately 400 eV, ascribed to N in EDTA. Meanwhile, the Sn 3d peaks of EDTA-SnO₂ shift to higher binding energy position by approximately 0.13 eV comparing to the pristine SnO2, indicating that EDTA is bound to the SnO₂ instead of simple mixing. The specific process of the reaction is shown in Fig. S1. The surface potentials of SnO2 and EDTA-SnO2 were tested by KPFM and the Fermi levels were calculated. The details of calculation are described in Note S1. As is shown in Fig. 2, the Fermi level of EDTA-SnO₂ is -4.10 eV, which is 0.08 eV higher than that of SnO₂. Fig. 3 presents the energy band structures of different HSCs (more details are shown in Note S2) (Amorim et al., 2012; Ke et al., 2015; Melskens et al., 2018). SnO₂ has an energy band structure matched well with silicon. The extremely deep valence band of SnO_2 can block the transmission of holes from silicon to Ag rear electrode effectively, thus helping to separate the photogenerated electrons and holes.

In order to evaluate the impedance to carrier transmission of Si/Ag,

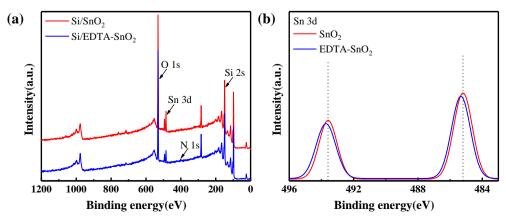


Fig. 1. (a) XPS spectra of SnO₂ and EDTA-SnO₂ on silicon. (b) XPS spectra of Sn 3d of SnO₂ and EDTA-SnO₂ on silicon.

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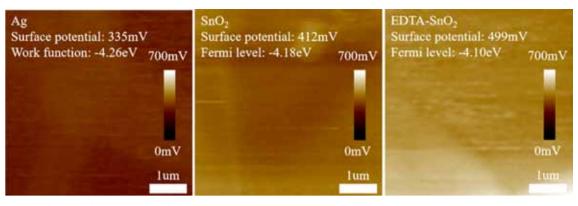


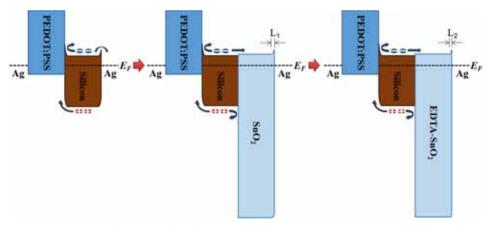
Fig. 2. Surface potentials and corresponding Fermi levels or work function of Ag, SnO2 and EDTA-SnO2 on silicon.

Si/SnO₂/Ag and Si/EDTA-SnO₂/Ag rear structures, we tested the resistance between two Ag rear electrodes of different structures. J-V curves of different rear structures and schematic diagram of the test structure are shown in Fig. 4. Due to the Fermi level pinning effect, high Schottky barrier (Φ_{B1} 0.56–0.76 eV) exhibits in Si/Ag contact (Huang and Hai, 1979). J-V curve of Si/Ag structure shows obvious rectification effect and high contact resistance. It is reported that there is a lower Schottky barrier (Φ_{B2} 0.31 eV) exhibits in SnO₂/metal contact (Amorim et al., 2012). Therefore, the lower resistance of Si/SnO₂/Ag structure than Si/Ag structure can be attributed to the lower barrier height caused the Fermi level depinning effect of SnO2 on silicon surface. However, since SnO₂ prepared by spin coating is not uniformly coated on silicon surface (stronger Si 2s peak than Sn 3d peak found in Fig. 1(a)), Si/SnO₂/Ag structure still shows strong rectification effect. Moreover, Si/EDTA-SnO₂/Ag shows lowest contact resistance. According to the characteristics of n-type semiconductor/metal contacts, higher Fermi level of n-type semiconductor will be beneficial to reduce the width of the Schottky barrier caused by Fermi level pinning, thus facilitating the transmission of carriers. Therefore, EDTA-SnO2 with higher Fermi level leads to lower contact resistivity comparing to SnO₂. The minority carrier lifetimes of Si, Si/SnO2 and Si/EDTA-SnO2 structures are all about 10 µs (more details are shown in Fig. S2), which means that carrier recombination rates on the silicon rear surfaces of all structures are very high. Therefore, although it is reported that SnO2 will form Si-O-Sn bonds with silicon atoms (Chen et al., 2018), SnO2 or EDTA-SnO₂ ESCs cannot passivate silicon surface effectively.

We prepared $\rm SnO_2$ and EDTA-SnO_2 ESCs in Si/PEDOS:PSS HSCs and tested the corresponding photovoltaic performance. Fig. 5 shows the light J-V curves and EQE curves of the HSCs with different ESCs. Table 1 shows the photovoltaic parameters of different HSCs. HSCs without ESCs exhibits a moderate η of 9.31%, which is mainly limited by a relatively low V_{oc} (517 mV) and FF (64.5%). The V_{oc} of HSCs with

SnO₂-based ESCs is markedly higher than that without ESCs, which is similar to the result that the V_{oc} was significantly increased after the insertion of TiOx ESCs in p-n junction silicon solar cell (Yang et al., 2016). Higher V_{oc} means that photogenerated electrons and holes in the HSCs with ESCs can be better separated when voltage applied, which is just the original intention of inserting ESCs. The V_{oc} of HSCs with EDTA-SnO₂ ESCs is 5 mV higher than SnO₂ ESC, which is caused by the closer Fermi level of EDTA-SnO₂ to silicon conduction band. It is believed that higher Fermi level of ESC will be beneficial to extract electrons, thus enhancing the V_{oc} of solar cells (Snaith et al., 2010; Yang et al., 2018). In addition, due to the lower barrier width caused by higher Fermi level of EDTA-SnO₂, Si/EDTA-SnO₂/Ag exhibits lowest contact resistance. Therefore, due to better electron extraction ability and lower contact resistance, the FF of HSCs with EDTA-SnO2 ESCs reaches 69.4% comparing to 65.8% of HSCs with SnO2 ESCs. However, even though ESCs can help to separate photogenerated electrons and holes, the J_{sc} of the HSCs with SnO2-based ESCs is just slightly higher than HSCs without ESCs. The EQE curves of HSCs with different rear structures are basically the same in the range of 550-1100 nm. The main reason is that SnO₂ and EDTA-SnO₂ cannot effectively passivate the dangling bonds on the silicon surface. Due to the lower barrier in the rear contact region caused by the insertion of ESCs in HSCs, the utilization of highenergy photons has been slightly improved.

As is known to all, a thin (about $1-2\,\mathrm{nm}$) SiO_x film can chemically passivate the dangling bonds on the silicon surface and reduce the carrier recombination rate, which still allow carriers to transport between layers (Feldmann et al., 2014a, 2014b; Moldovan et al., 2015). Therefore, we place the silicon wafer under $\mathrm{UV/O_3}$ photo-oxidation treatment for 4 min to produce a thin SiO_x passivation film with appropriate thickness, which is efficient and cheap (Moldovan et al., 2015). By preparing the SiO_x passivation film, the hydrophilicity of the silicon surface is significantly enhanced, which is beneficial to spin



 $\textbf{Fig. 3.} \ \, \textbf{Energy band structures of different Si/PEDOT:PSS HSCs.}$

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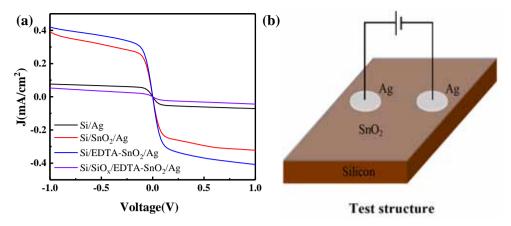


Fig. 4. (a) J-V curves of different rear structures. (b) Test structure of the resistance of different rear structures.

Table 1

coating more uniform EDTA-SnO₂. We have prepared the Si/SiO_y/ EDTA-SnO₂/Ag structure and evaluate the contact resistivity. The *J-V* of Si/SiO_x/EDTA-SnO₂/Ag structure is shown in Fig. 4. Through comparison, we find that the resistance of Si/SiO_x/EDTA-SnO₂/Ag is higher than Si/EDTA-SnO₂/Ag, which is caused by the high resistance of the SiO_x film. However, the rectification characteristic of the *J-V* curve of Si/SiO_x/EDTA-SnO₂/Ag is obviously weakened, mainly caused by more uniform EDTA-SnO₂ coating (more details are shown in Fig. S3). After oxidation passivation, the Si/SiOx/EDTA-SnO2 structure exhibits much higher minority carrier lifetime of 13.4 µs than other structures (about 10 µs), which indicates the lowest carrier recombination rate on the silicon surface. Fig. 5 presents the light J-V curve and EQE curve of HSCs with EDTA-SnO₂/SiO_x bilayer ESCs. Photovoltaic parameters of the HSCs are shown in Table 1. Due to more EDTA-SnO2 coating and lower carrier recombination rate, HSCs with EDTA-SnO₂/SiO_x bilayer ESCs exhibit higher η than HSCs with EDTA-SnO₂ ESCs, which is mainly caused by the increase in V_{oc} and FF by 13 mV and 1.8% respectively. We note that increased contact resistance caused by a thin SiO_x film will not hinder the transmission markedly. Moreover, the EQE curve of the HSC with EDTA-SnO₂/SiO_x bilayer ESC is slightly improved in the range of 300-950 nm. The further increase of J_{sc} is ascribed to Fermi level depinning effect of EDTA-SnO2 and reduction of carrier recombination rate on the silicon surface. After optimization, the planar Si/PEDOT:PSS HSCs with EDTA-SnO₂/SiO_x bilayer ESCs achieves a higher η of 11.52% than 10.13% of that with SnO₂ ESCs.

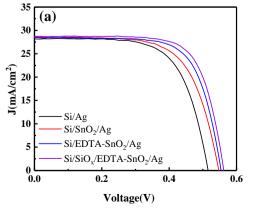
4. Conclusion

Due to its suitable energy band structure, SnO2 ESC can help to separate photogenerated electrons and holes in Si/PEDOT:PSS HSC.

Photovoltaic parameters of the HSCs with different rear structures.

Rear structure of HSC	V_{oc} (mV)	J_{sc} (mA/cm ²)	FF (%)	η (%)
Si/Ag Si/SnO ₂ /Ag Si/EDTA-SnO ₂ /Ag Si/SiO _x /EDTA-SnO ₂ /Ag	544 549 ± 4	28.0 ± 0.8 28.3 ± 0.4 28.2 ± 0.7 28.8 ± 0.1	65.8 ± 0.7 69.4 ± 2.0	9.31 ± 0.49 10.13 ± 0.27 10.75 ± 0.18 11.52 ± 0.08

Moreover, SnO₂ ESC can eliminate the high Schottky barrier caused by Si/Ag contact through Fermi level depinning effect. However, the performance of corresponding HSCs will be limited by the unsatisfactory electron extraction ability and limited surface passivation effect of SnO₂ ESCs. Through doping of EDTA, the Fermi level of EDTA-SnO₂ is closer to the conduction band of silicon than SnO₂, which will benefit the transmission of photogenerated electrons and enhance the Voc of HSCs. Doping of EDTA also endues Si/EDTA-SnO2/Ag lower contact resistivity than Si/SnO₂/Ag. Due to better separation of carriers and lower contact resistance, the FF of HSCs with EDTA-SnO₂ ESCs reaches 69.4% comparing to 65.8% of HSCs with SnO₂ ESCs. However, neither SnO2 nor EDTA-SnO2 can provide effective passivation for silicon surface. Therefore, EDTA-SnO₂/SiO_x bilayer ESC is prepared by combining a EDTA-SnO₂ layer and a thin SiO_x film grown on silicon surface. It is found that although SiO_x film increases the contact resistance of the rear region, the reduced carrier recombination rate and more uniform distributed EDTA-SnO2 enables devices to have better performance. The planar Si/PEDOT:PSS HSCs with EDTA-SnO₂/SiO_x bilayer ESCs exhibit a higher η of 11.52% than 10.13% of that with SnO_2 ESCs, mainly caused by the improved V_{oc} and FF.



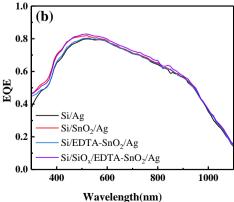


Fig. 5. (a) Light J-V curves of HSCs with different rear structures. (b) EQE curves of HSCs with different rear structures.

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Appendix A. Supplementary material

Supplementary data to this article can be found online at https://doi.org/10.1016/j.solener.2019.09.077.

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